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# Coulomb corrections and thermo-conductivity of a dense plasma

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#### **Abstract**

We point out that confusion sometimes arises when using a chemical potential in plasma with Coulomb interactions. The results of our consideration are applied to the discussion of nuclear reactions screening. Finally, we present a transparent derivation of the thermal conductivity coefficient of a degenerate electron gas.

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#### 1. Introduction

This paper considers some questions that are important when nuclear flames in a dense plasma are considered. These questions can be applied to many systems, for example, to supernovae of type Ia. If one considers papers dealing with nuclear statistical equilibrium (NSE) or with nuclear reactions screening, one can note that authors use different definitions of chemical potential and different expressions for Coulomb corrections. Some of these definitions are general and some are accurate only in particular cases. Inaccurate definitions can sometimes lead to the wrong conclusions and misleading understanding of a physical situation. The aim of this paper is to arrange all definitions and basic equations into a physical perspective and to discuss some paradoxes arising from inaccurate definitions. The second part of the paper deals with kinetic properties of matter in white dwarfs. The question was considered in the 1960–1970s, [1, 2] and references therein. One of the most important parameters for supernovae Ia is thermo-conductivity of a degenerate electron gas. A method for calculating the thermal conductivity coefficient was described in [1], but the final expression was not obtained; whereas in [2] a correct final expression was obtained but derivation is not so rigorous. There are some improvements in [3–5]. Here we present a straightforward and more transparent derivation, based on the ideas from [1].

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#### 2. Chemical potential

A general definition of a chemical potential  $\mu$  is contained in the differential of internal energy E.

$$dE = T dS - P dV + \mu dN. \tag{1}$$

We use standard notations here and  $\mu$  dN may be a sum for several components:  $\mu$  dN  $\rightarrow \sum_i \mu_i dN_i$ . The differential of the Helmholtz free energy  $F \equiv E - TS$  is written as follows:

$$dF = -S dT - P dV + \mu dN, \qquad (2)$$

and thus it means that

$$\mu = \left(\frac{\partial F}{\partial N}\right)_{T,V}.\tag{3}$$

Let us consider the Coulomb free energy of a set of  $N_i$  nuclides;

$$F_{\text{nuc}} = \sum_{i} N_i \Phi_i(T, n_k). \tag{4}$$

Here  $\Phi_i$  is the free energy per nucleus i and it may depend not only on the concentration,

$$n_i = \frac{N_i}{V},\tag{5}$$

of the *i*th nucleus but also on concentrations of some other nuclei  $n_k$ . We have by definition

$$\mu_i = \frac{\partial F_{\text{nuc}}}{\partial N_i} \bigg|_{T,V} = \Phi_i + \sum_k n_k \frac{\partial \Phi_k}{\partial n_i} \bigg|_{T}.$$
 (6)

Then we find, for example, the pressure of nuclide gas:

$$P_{\text{nuc}} = -\frac{\partial F_{\text{nuc}}}{\partial V} \bigg|_{T,\{N_k\}} = \sum_{k} \mu_k n_k - F_{\text{nuc}}/V, \tag{7}$$

which is consistent with the definition of the grand thermodynamic potential,  $\Omega = -PV$ . A convenient parameter of plasma non-ideality is

$$\Gamma_i \equiv \frac{1}{k_{\rm B}T} \left( \frac{e^2 Z_i^2}{a_i} \right) = \frac{e^2 Z_i^2}{k_{\rm B}T} \left( \frac{4\pi}{3} n_i \right)^{1/3}.$$
 (8)

According to accurate Monte-Carlo (MC) simulations, a very good theoretical model for Coulomb corrections in a neutral one-component plasma is Wigner–Seitz approximation (in the first order of  $\Gamma$ ), see references, e.g. in [6]. In this approximation, the correction to the free energy of a system is

$$\Delta F^C = -0.9 N_i \Gamma_i k_B T, \tag{9}$$

where  $a_i$  is an average distance between ions of type i,

$$a_i = \left(\frac{3}{4\pi} \frac{V}{N_i}\right)^{1/3}.\tag{10}$$

Such a definition of  $a_i$  is often used (see e.g. [7]), but we point out in advance of a full discussion that this is correct only for one-component plasma.

A good correspondence of this simple model with MC simulation one may find, e.g. in the results by [8], where the main correction to free energy is approximated by  $f = a\Gamma + \cdots$  and a = -0.899172.

So in the case of the Wigner–Seitz approximation, the Coulomb correction to chemical potential from equations (3) and (9) is

$$\Delta \mu^C = -1.2 \frac{Z_i^2 e^2}{a_i} = -1.2 \Gamma_i k_B T, \tag{11}$$

the result which has a coefficient that is 4/3 times larger than the Wigner–Seitz's 0.9 because  $a_i$  depends on  $N_i$  too. Here and below we will focus our attention on the numerical coefficient in the correction; thus, the coefficient is very important.

Sometimes  $a_i$  is introduced in another way, through electron density

$$a_i = \left(\frac{3}{4\pi} \frac{Z_i}{n_e}\right)^{1/3},\tag{12}$$

and the resulting expression for the Coulomb corrections is

$$\Delta F^C = -0.9 N_i Z_i^{5/3} e^2 \left(\frac{4\pi}{3} n_e\right)^{1/3} . \tag{13}$$

At first glance such an expression leads to the coefficient 0.9 in (11), but in this case  $n_e$  depends on the number of ions. If we take out one ion from our system then, for preserving electro-neutrality, we should take out Z electrons; hence,  $n_e$  changes so that we obtain the correct answer (11).

These considerations have direct implications for NSE which is essentially reduced to a Saha-like equation for abundances of nuclides. A correct form of these equations should be derived from correct expressions for chemical potentials in equilibrium. However, it is sometimes also used in another approach, when Coulomb corrections (with the coefficient 0.9) are introduced directly to the binding energies of reacting nuclei. This approach also seems natural, at least while nuclides form an almost ideal gas (see, e.g., a widely cited paper [9]), but the results for nuclear abundances will differ from those obtained with the correction (11) for chemical potential.

We can derive the result (11) from a more general expression. If  $Q_i$  denotes the Coulomb correction to the free energy and has the functional form

$$Q_i \equiv k_B T f_0(\Gamma_i), \tag{14}$$

then we find the Coulomb contribution to the chemical potential as

$$\mu_i^{\text{Coul}} = k_B T \left[ f_0(\Gamma_i) + \frac{1}{3} \frac{\partial f_0}{\partial \ln \Gamma_i} \right]; \tag{15}$$

this agrees with (11) in case of the Wigner–Seitz approximation. But we want to point out that it contradicts equation (17) of [9] often used in NSE studies. Our expression (15) does not coincide also with equation (97) of the famous review by [10]. It was found in [10] that  $\mu_i = k_B T f_0(\Gamma_i)$ , but this expression leads to zero correction to  $P_{\text{nuc}}$  in equation (7). On the other hand, our equation (15) gives the expected correction to  $P_{\text{nuc}}$  after substitution into equation (7).

It is a very important point in our reasoning, so let us consider it carefully. As will be shown in section 3, a more appropriate expression for Coulomb corrections is (13) and not (8)–(9). But if we use the assumption of electro-neutrality, the numbers of electrons and ions are directly related. It is correct when we consider our system on a timescale that is much larger than plasma frequency (like in NSE)—so we cannot separate electrons and ions. In this case, the chemical potential is defined as

$$\mu = \left. \frac{\partial F}{\partial N_i} \right|_{T, V, N_{\text{ions, electro-neutr.}}},\tag{16}$$

and for a one-component mixture, the answer is (11). Here the full correction could be ascribed to ions if electrons are considered as background or to electrons if ions are background. But when we consider kinetic properties of our medium, we study it on timescales when electroneutrality could be violated—so we should consider electrons and ions separately, like the corrections

$$\Delta \mu_{i}^{C} = \frac{\partial \Delta F_{C}}{\partial N_{i}} \Big|_{T,V,N_{e}} = -0.9 Z_{i}^{5/3} e^{2} \left(\frac{4\pi}{3} \frac{N_{e}}{V}\right)^{1/3},$$

$$\Delta \mu_{e}^{C} = \frac{\partial \Delta F_{C}}{\partial N_{e}} \Big|_{T,V,N_{i}} = -0.3 \frac{N_{i}}{N_{e}} Z_{i}^{5/3} e^{2} \left(\frac{4\pi}{3} \frac{N_{e}}{V}\right)^{1/3};$$
(17)

these corrections should be applied to (46) below (there, in a kinetic case, all thermodynamic quantities have an approximate meaning, and have to be carefully defined similar to that in [11]). Here we should emphasize one important point. Coulomb corrections in dense plasma arise due to electron—ion interactions because all ions are screened by electron clouds. The notation  $F_{ii}$  that is used in many works is a bit misleading because it is not a pure ion—ion interaction.

In the book by [12], the authors rely mostly on free energy F instead of  $\mu_i$ . But in many cases correct knowledge of  $\mu$  is required. Actually, the expression which is equivalent to (15) appeared already in [13]. In the paper [14], the result for the  $\mu$  coefficient is close to 1.2, but is was obtained using Monte-Carlo simulation and rough expansion. However, the argument on the role of the chemical potentials and debates on approaches to them still persists. For example, in [15] the difference of their use of chemical potentials have been discussed with that of [14]. Essentially, in [15] it was found that the Coulomb correction found in [14] for a classical plasma is applicable to quantum plasma as well. (See also [16] for a detailed review on modern perspective on Coulomb systems.)

Since there is some controversy on the form of Coulomb corrections in the literature, in the paper [17] the results for the Coulomb correction from equation (14), as was taken by [9], with the effect of the additional term in equation (15), have been compared. The effect of the new term in (15) on the proton fraction (which is very important in collapse computations) is visible only at highest density but still very small. A bit more pronounced effect is obvious for the average mass of heavy nuclei again at highest density when the NSE model is near the border of applicability. However, a more detailed study is needed here along the lines undertaken recently by [18].

#### 3. Screening of nuclear reactions

A very important application of the previous reasoning is in the field of a classical nuclear burning screening. Consider the regime of strong screening  $\Gamma \geqslant 1$ . According to [14] the expression for the screening factor in this case can be written as

$$\exp(H_{12}(0)) = \exp(-\{\mu(Z_1 + Z_2) - \mu(Z_1) - \mu(Z_2)\}/T). \tag{18}$$

Here and below  $k_B \equiv 1$ . So theoretically, using the above definition of a chemical potential, the screening factor is

$$H_{12}(0) = 1.2(\Gamma_{Z_1 + Z_2} - \Gamma_{Z_1} - \Gamma_{Z_2}). \tag{19}$$

In order not to create confusion, let us consider a commonly examined situation when  $Z_1 = Z_2 = Z$ . We obtain in the leading order (here we will discuss only the leading order of corrections)

$$H_{12}(0) = 1.410\Gamma_Z \tag{20}$$

(here  $1.410 = 1.2(2^{5/3} - 2)$ ). On the other hand, if we consider the definition of the screening factor using free energy,

$$H_{12}(0) = -(F(2Z) - 2F(Z))/T, \tag{21}$$

and using (9) obtain

$$H_{12}(0) = 0.9(\Gamma_{2Z} - 2\Gamma_{Z}) = 1.057\Gamma_{Z}.$$
(22)

In the case of  $\Gamma \gg 1$  the difference in (19) and (22) can be extremely high. In some works the chemical potential has a coefficient 0.9 and so using two definitions of  $H_{12}$  the answer (22) is obtained.

First of all, let us consider the physical sense of a chemical potential. Equation (18) tells us that we take out from our system two ions with a charge Z and put back one ion with a charge 2Z. From the chemical point of view, ions with Z and 2Z are different elements and the average distance between Z-ions changes. So if we recall that the coefficient 1.2 arises from the change of  $a_i$ , then (22) is incorrect. But for finding a proper screening correction, one should stand on electromagnetic force point of view: these are the elements of the same nature—and there should be only one distance between reactants.

At this point a reader can object in the following way: a nuclear reaction is a fast process, but the effect we are discussing now exists because of a global (on scales of our system) variation of the average distance between ions: this is a slow process. Can it really change the repulsion potential—a barrier, our ion is penetrating through? Theory of such corrections rely on a statistical averaging [14, 19] and operate with average quantities, so in such a theory we should consider these effects.

On the other hand, if we consider the definition of free energy using  $n_e$  (13), in nuclear reactions, the electron number density  $n_e$  stays constant, unlike the situation when we define chemical potential and take one ion with its electron cloud from our system. The average distance  $a_i$  does not change in these terms, and then the coefficient 0.9 should stay in (19). We see that there is some paradox in this situation: one reasoning leads to 0.9, another to 1.2.

Let us resolve the paradox. Expression (22) used by scientific community is correct. The reason of the misunderstanding is the definition of  $a_i$  which is sometimes incorrect. What is the Wigner–Seitz model? This model tells us that the most significant Coulomb corrections are due to the interaction of ions with electrons (and defining these correction through the average distance between ions may result in a confusion of the reader). Electron gas is uniformly spread over the whole volume (we omit such effects as electron polarization etc) and we should define  $a_i$  for an ion  $Z_i$  as a radius around the ion, which contains electron charge  $(-Z_i)$ :

$$a_i \equiv \left(\frac{3}{4\pi} \frac{Z_i}{n_o}\right)^{1/3}.\tag{23}$$

Here, in general,  $a_i$  is not the average distance between ions: this would be true only if we had a one-component mixture—what is undoubtedly wrong in nuclear reactions, our mixture after reaction is multicomponent. So the correct definition of  $a_i$  is a radius of electron cloud with charge  $(-Z_i)$ . Such an electron coat screens nuclear repulsion potential and one ion from another—there is no direct Coulomb interaction between ions. As we have mentioned above,  $n_e$  is constant, so the correct leading order in screening is

$$H_{12}(0) = 1.057\Gamma_Z. \tag{24}$$

Finally we could write the full classical correction from MC simulation with equal charges (see e.g. [8]):

$$H_{12}(0) = 1.056299\Gamma_Z + 1.039957\Gamma_Z^{0.323064} - 0.545823\ln\Gamma_Z - 1.036319.$$
 (25)

Let us emphasize one moment. When defining chemical potential, we should obey the electroneutrality law and take one ion from the system with its electron cloud. And in nuclear reactions nothing is taken from system, so the chemical potential has another physical meaning, and the correct definition is done by free energy (21). A very bright demonstration of this point is the famous review by Ichimaru [5] with its section C on 'chemical potentials': only free energies are discussed, and  $\mu$  symbol does not appear there.

#### 4. Thermal conductivity

It is commonly supposed that write dwarfs (WD) are progenitors for supernovae Ia. On the initial stage of explosion, thermo-conductivity plays a significant role as it drives the flame front of thermonuclear burning [20]. Typical parameters of matter in presupernovae WD are  $T \sim 10^9$  K and  $\rho \sim 10^9$  g cm<sup>-3</sup>. Chemical composition is a mixture  $^{12}\text{C+}^{16}\text{O}$ . For these conditions the electron gas is relativistic and degenerate  $\epsilon_F \gg k_B T$ ,  $p_F \gg m_e c$ , and the parameter of non-ideality (8)  $\Gamma \sim 1$ .

Let us consider an electron gas which is described by its distribution function in a phase space  $f(\mathbf{p}, \mathbf{x}, t)$ . Boltzmann equation for evolution of such a function is the following (in this section we will use units in which  $c = \hbar = k_B = 1$  and in the end we will restore dimensionality of variables):

$$\frac{\mathrm{d}f}{\mathrm{d}t} = \mathrm{St}f = -\int w(\mathbf{p}, \mathbf{p}')[f(1-f') - f'(1-f)] \frac{2\mathrm{d}^3 p'}{(2\pi)^3}.$$
 (26)

Here we assume that the scattering of electrons occurs only on ions, and the distribution function of the latter does not change. Under our conditions the ions are non-degenerate, non-relativistic and have large mass in comparison with electrons' mass; they can be treated as being at rest. An approach which considers the scattering on one ion (separately from other ions) is incorrect when  $\Gamma\gg 1$  because of the correlation in ion positions. Thus, we shall consider a situation when  $\Gamma\leqslant 1$  and suppose that our answer has a right limit  $\Gamma\to 1$  (finally we want the answer for  $\Gamma\sim 1$ ). This is true because when  $\Gamma\ll 1$ , we have an ideal gas behavior of ions. The case  $\Gamma\sim 1$  is called Coulomb liquid: there is no significant correlation in positions and full correlation sets in the solid state when  $\Gamma\sim 170$  (see [21–23]).

Cross-section of scattering on an ion at rest (with potential  $A_0(r) = Ze/r$ ) is written as follows [24] (we will use a short definition  $d\mathbf{p}' \equiv 2d^3p'/(2\pi)^3$ ):

$$\frac{\mathrm{d}\sigma}{\mathrm{d}\mathbf{p}'} = \pi e^2 \delta(\epsilon_{p'} - \epsilon_p) |A_0(\mathbf{p}' - \mathbf{p})|^2 \left(1 - \frac{\mathbf{q}^2}{4\epsilon_p^2}\right) \frac{\epsilon_p}{|\mathbf{p}|},\tag{27}$$

where  $A_0(\mathbf{q}) = 4\pi Ze/\mathbf{q}^2$ . The  $\delta$ -function tells us that the scattering is elastic. The comparison of e-i and e-e cross-sections gives  $\sigma_{\rm ei}/\sigma_{\rm ee} \sim Z^2 \gg 1$ , see e.g. [24], so we omit the e-e scattering here.

As usual, we consider a small deviation from the equilibrium distribution function:

$$f_p = f_p^0 - \frac{\partial f_p^0}{\partial \epsilon_p} \phi_p = f_p^0 + \frac{1}{T} f_p^0 (1 - f_p^0) \phi_p, \qquad f_p^0 = \frac{1}{e^{(\epsilon_p - \mu)/T} + 1}.$$
 (28)

So Boltzmann equation in the case of thermo-conductivity looks like

$$-\frac{\partial f^0}{\partial \epsilon} \frac{\epsilon - w}{T} (\mathbf{v} \nabla) T = -\frac{1}{T} \int w(\mathbf{p}, \mathbf{p}') (\phi - \phi') f^0 (1 - f^0) \frac{2d^3 p'}{(2\pi)^3}.$$
 (29)

In the case of a degenerate electron gas,  $sT \ll \mu$ , so we could substitute  $\mu$  instead of w. Let us define the abbreviation  $\Gamma_{p \to p'} \equiv w(\mathbf{p}, \mathbf{p}') f^0 (1 - f^0)$ . After multiplication on  $\phi$  and integration over the whole p, we can reduce the Boltzmann equation to the form

$$\frac{j_E^2}{\kappa T} = \frac{1}{2T} \int (\phi - \phi')^2 \Gamma_{p \to p'} \frac{2d^3 p}{(2\pi)^3} \frac{2d^3 p'}{(2\pi)^3},\tag{30}$$

where the energy flux is defined as

$$\mathbf{j}_{\mathbf{E}} = \int (f - f^0) \mathbf{v} (\epsilon - \mu) \frac{2d^3 p}{(2\pi)^3},\tag{31}$$

and the thermo-conductivity  $\kappa$  is defined in the following way:

$$\mathbf{j}_{\mathbf{E}} = -\kappa \nabla T. \tag{32}$$

Equation (30) has exact physical meaning—it can be shown (see [25]) that the rhs of (30) is a rate of entropy generation by particles' scattering. On the lhs is the entropy generation rate through conductivity. The following method of coefficient determination is good—we could substitute arbitrary functions  $\phi$  and get answers for  $\kappa$ . There exists a theorem that the maximum value of  $\kappa$  for different  $\phi$  is mathematically exact.

This is a simple case, namely, isotropic. There is a limited set of quantities in our task; that is why we could guess the answer

$$\phi = A(\epsilon_p - \mu) p_\alpha \partial_\alpha T. \tag{33}$$

The previous expression can be simply explained: it should be proportional to the temperature gradient, the only physical quantity that can create a scalar with combination with gradient is a momentum. The factor  $(\epsilon_p - \mu)$  was introduced because all macroscopic parameters should be defined through  $f_0$  but not through the correction proportional to  $\phi$ . The factor A is a dimensional constant, but it will be cancelled out immediately after the insertion of  $\phi$  into (30), so we omit it now.

Let us consider the integration

$$\mathbf{j}_{\mathbf{E}} = \frac{1}{T} \int (\epsilon_p - \mu) \mathbf{v} f_p^0 (1 - f_p^0) \phi \, \mathrm{d}\mathbf{p}. \tag{34}$$

Choosing for definiteness  $\partial_{\alpha}T=(0,0,\nabla T)$  and using the fact of high degeneracy of electrons (therefore,  $\int_0^{\infty}\psi(\epsilon)\frac{\partial f_0}{\partial \epsilon}\,\mathrm{d}\epsilon=-\psi(\mu)-\frac{\pi^2}{6}T^2\psi''(\mu)-\frac{7\pi^4}{360}T^4\psi''''(\mu)+\cdots$ ), we obtain

$$j_{E_z} = \frac{1}{9} (\nabla T) T^2 \left( \mu^2 - m_e^2 \right)^{3/2}. \tag{35}$$

The rhs of equation (30):

RHS = 
$$\int (\phi - \phi')^2 \Gamma_{p \to p'} \frac{2d^3 p}{(2\pi)^3} \frac{2d^3 p'}{(2\pi)^3},$$
 (36)

after all substitutions is

RHS = 
$$\int (\epsilon_p - \mu)^2 q_\alpha (\partial_\alpha T) q_\beta (\partial_\beta T) \pi e^2 \delta(\epsilon - \epsilon') |A_0|^2 \left( 1 - \frac{\mathbf{q}^2}{4\epsilon_p^2} \right) \frac{\epsilon_p}{p}$$
$$\cdot f_p^0 \left( 1 - f_p^0 \right) n_i v \frac{2d^3 p}{(2\pi)^3} \frac{2d^3 p'}{(2\pi)^3}. \tag{37}$$

Doing the same integration,

$$RHS = \frac{1}{18\pi} e^2 n_i T^3 \mu \left(\mu^2 - m_e^2\right)^{1/2} (\nabla T)^2 \frac{\epsilon}{p} \int_0^{2p} q^3 |A_0(q)|^2 \left(1 - \frac{\mathbf{q}^2}{4\mu^2}\right) dq.$$
 (38)

Let us denote Coulomb logarithm (recalling that  $A_0(q) = 4\pi Ze/q^2 = 4\pi Ze\varphi(q)$ )

$$\Lambda_{\rm ei} = \int_0^{2p} q^3 |\varphi(q)|^2 \left(1 - \frac{\mathbf{q}^2}{4\mu^2}\right) \mathrm{d}q; \tag{39}$$

RHS = 
$$\frac{8\pi}{9} Z^2 e^4 n_i T^3 \mu (\mu^2 - m_e^2)^{1/2} (\nabla T)^2 \frac{\epsilon}{p} \Lambda_{ei}$$
. (40)

After the substitution of (35) and (40) in (30), and defining  $v_F \equiv p/\epsilon$ , we obtain

$$\kappa = \frac{1}{36\pi} \frac{T(\mu^2 - m_e^2)^{5/2} v_F}{Z^2 e^4 n_i \mu \Lambda_{ei}},\tag{41}$$

recovering dimension

$$\kappa = \frac{1}{36\pi} \frac{k_B^2 T \left(\mu^2 - m_e^2 c^4\right)^{5/2} v_F}{(\hbar c)^3 Z^2 e^4 n_i \mu \Lambda_{ei}}.$$
(42)

This is the same result as calculated in [2].

Last uncomputed quantity is the Coulomb logarithm. Because of the quantum nature of our scattering (see [11]),

$$\frac{|ee'|}{\hbar v} \approx \frac{Ze^2}{\hbar c} = Z\alpha \ll 1,\tag{43}$$

then the minimal momentum transferred is  $q_{\min} = \hbar/l$ , where l is the typical length of interaction. Then (39) can be rewritten as

$$L_{\rm ei} = \ln\left(\frac{2p_e l}{\hbar}\right) - \frac{v_F^2}{2c^2}.\tag{44}$$

We should define l—maximum length with interaction. Recalling reasoning from the previous paragraph, we could put  $l = a_i$ , so in this case the Coulomb logarithm:

$$L_{ei} = \ln((18\pi)^{1/3}Z^{1/3}) - \frac{v_F^2}{2c^2} = \ln(3.838Z^{1/3}) - \frac{v_F^2}{2c^2}.$$
 (45)

In hydrodynamics there are general expressions for mass and heat fluxes (see, e.g. [26]):

$$\mathbf{i} = -\alpha \nabla \mu - \beta \nabla T, \qquad \mathbf{q} = -\delta \nabla \mu - \gamma \nabla T.$$
 (46)

Here we use a definition of  $\mathbf{q}$  with subtraction of  $\mu \mathbf{i}$ . Let us restore terms with  $\nabla \mu$  in the kinetic equation:

$$-\frac{\partial f_0}{\partial \epsilon} \frac{\epsilon - \mu}{T} (\mathbf{v} \nabla) T + \frac{\partial f_0}{\partial \epsilon} (\mathbf{v} \nabla) \mu = \int w(\mathbf{p}, \mathbf{p}') (\phi - \phi') \frac{\partial f_0}{\partial \epsilon} d\mathbf{p}'. \tag{47}$$

Like above, we look for the solution in the form

$$\phi = A(\epsilon - \mu)(\mathbf{p}\nabla)T + B(\mathbf{p}\nabla)\mu. \tag{48}$$

And using the same techniques, we obtain

$$\phi = A(\epsilon - \mu)(\mathbf{p}\nabla)T + AT(\mathbf{p}\nabla)\mu, \qquad A = -\frac{1}{4\pi} \frac{(\mu^2 - m_e^2)v_F}{Z^2 e^4 n_i \mu \Lambda_{ei} T}.$$
 (49)

Based on the definitions

$$\mathbf{i} = -\int \mathbf{v} \frac{\partial f_0}{\partial \epsilon} \phi \, d\mathbf{p}', \qquad \mathbf{q} = -\int (\epsilon - \mu) \mathbf{v} \frac{\partial f_0}{\partial \epsilon} \phi \, d\mathbf{p}', \tag{50}$$

we obtain Onsager's result  $\delta = T\beta$  and in our case (of degenerate electrons)  $\alpha = -AT\frac{v_F^2}{3}\mu$ ;  $\beta = 0$ ;  $\delta = 0$ ;  $\gamma = \kappa$ .

At the end we can give some limits of applicability for our expression for  $\kappa$ . As we can recall, we have used

$$\int_0^\infty \psi(\epsilon) \frac{\partial f_0}{\partial \epsilon} \, \mathrm{d}\epsilon = -\psi(\mu) - \frac{\pi^2}{6} T^2 \psi''(\mu) - \frac{7\pi^4}{360} T^4 \psi''''(\mu) + \cdots$$
 (51)

in our calculations. We have omitted the third term and this is correct when

$$T < \sqrt{\frac{60}{7\pi^2} \frac{\psi''(\mu)}{\psi''''(\mu)}} = 0.16\mu,\tag{52}$$

where the last equality arises from our previous calculations. This is the validity criterion for our expression (42).

#### 5. Conclusions

This work presents methodical notes on Coulomb corrections in a non-ideal plasma. A quantity  $a_i$  that enters the corrections in different works is defined in a different manner, what can lead sometimes to inaccurate results. We emphasize that a correct definition of  $a_i$  is a radius of electron cloud with charge  $(-Z_i)$ :

$$a_i \equiv \left(\frac{3}{4\pi} \frac{Z_i}{n_e}\right)^{1/3}.\tag{53}$$

This quantity is used for the definition of free energy (9). When the relation (53) is used the physical sense of correction to F is clear: they arise due to the interaction of ions with electron cloud, and not ion—ion interactions.

An example of the quantity that depends significantly on such corrections is the classical screening of nuclear reactions. Sometimes chemical potentials are used for the definition of  $H_{12}(0)$ . We argue that a chemical potential does not reflect the essence of the matter: the interaction is fully electromagnetic in a screening process, and elements with different Z have qualitatively the same nature relative to these interactions, while no particles are taken from the system in nuclear reactions. A correct formula should be written in terms of free energy and should read

$$H_{12}(0) = -\left[F(N-2,1) - 2F(N,0)\right]/T,\tag{54}$$

where F(N-2, 1) is the free energy for N-2 ions with a charge Z and one ion with charge ZZ, while F(N, 0) is defined respectively for ions before reaction [19].

Finally, we give a straightforward derivation for a thermo-conductivity coefficient of a degenerate relativistic electron gas.

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